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definite proportions. law of multiple
proportions. atom. matter can't be
created or destroyed. chemical
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the same proporti.... Where two
compounds have two of the the same

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composed of positive and negative ions that are combined so that the numbers of positive and negative charges are equal. Formula Unit. the simplest collection of atoms from which an ionic compound's formula can be established.

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following questions in the space
provided. c c The molar mass of a gas at
STP is the density of that gas (a)

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Answers

multiplied by the mass of 1 mol. (b)
divided by the mass of 1 mol. nRT (c)
multiplied by 22.4 L. (d) divided by 22.4
L. For the expression $V =$ (a) increasing
P

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Bonding SECTION 3 IONIC BONDING AND IONIC COMPOUNDS SHORT ANSWER

Answer the following questions in the space provided. 1. _____ The notation for sodium chloride, NaCl, stands for one (a) formula unit.

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Periodic Law SECTION 1 SHORT ANSWER

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Answer the following questions in the space provided. 1. c In the modern periodic table, elements are ordered (a) according to decreasing atomic mass.

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nucleus of an isotope.

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REVIEW Gases Class SHORT ANSWER
Answer the following questions in the
space provided c c The molar mass of a
gas at STP is the

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